Draw Lewis Dot Diagrams and give structural formulas for each of the following:

A) H_2

B) OF_2

C) F_2O

D) N_2

E) CH_3F

F) S_8

G) CH_2Cl_2

H) $C_3H_6F_2$

I) $C_3H_2F_2$

J) CSO

Complete the table below for each molecule on the previous page

Molecule	# of lone pairs on the central atom	# of bonds to the central atom	