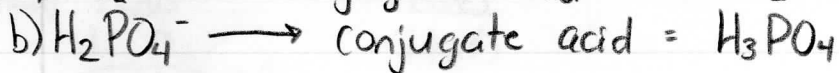
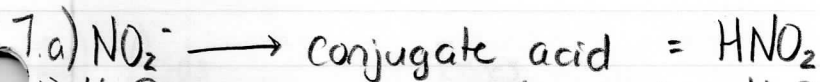
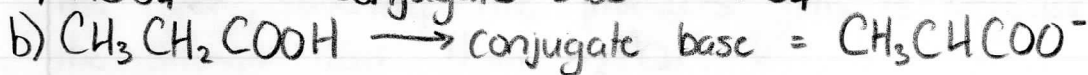
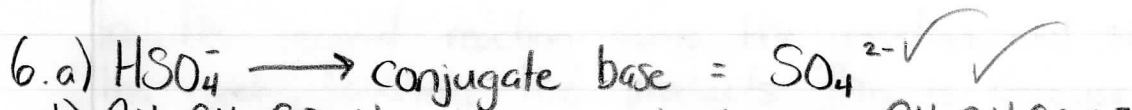


4. As well as the lack of evidence to support the formation of NH_4OH as an intermediate in the case of NH_3 acting as a base contributes to the flaws. (3)

5. A Brønsted-Lowry acid is considered a proton (H^+) donor in which reacts with a base. (2)

A Brønsted-Lowry base is considered a proton (H^+) acceptor in which reacts with an acid.



8. Water is considered as an "amphoteric" compound because it can react as both an acid or a base: (3)

