CLE 3221.3.3 Explore the mathematics of chemical formulas and equations.

SPI 3221.3.5 Convert among the following quantities of a substance: mass, number of moles, number of particles, molar volume at STP

## Chemistry Molar Volume at STP Practice Problems

Assume all of the following problems are at standard conditions.

- 1. What volume will 0.750 moles of nitrogen gas occupy?
- 2. What volume will 100. grams of oxygen gas occupy?
- 3. How many moles are contained in 2.500 liters of methane (CH<sub>4</sub>)?
- 4. Calculate the mass of 300. liters of dinitrogen monoxide.
- 5. What volume will 10. grams of ammonia occupy?

## Molar Volume at STP Practice Problems Answer Key

- 1.  $16.8 L N_2$
- 2. 70.0 L O<sub>2</sub>
- 3. 0.112 mol CH<sub>4</sub>
- 4. 589 g N<sub>2</sub>O
- 5. 13 L NH<sub>3</sub>