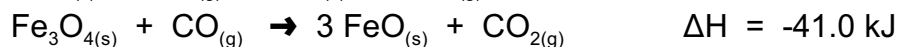
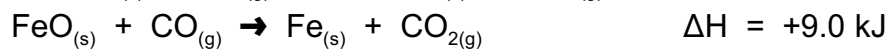
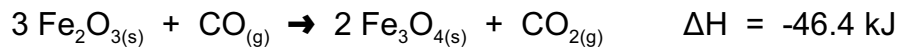
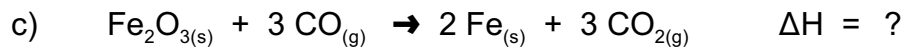
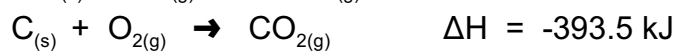
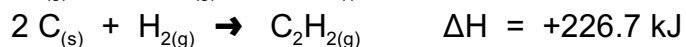
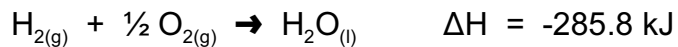
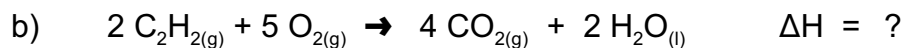
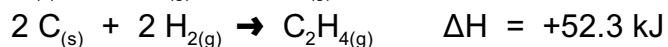
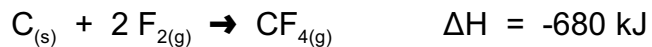
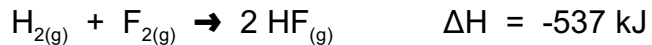


Hess's Law

1. Use the equations provided to determine the missing heat of reaction.



2. Use standard molar heats of formation on p. 848 to calculate the enthalpy change for:

